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Gaskell Thermodynamics Solutions Manual

thermodynamics has much wider applicability. In material science, one is normally not that interested in heat and work, but interested more the state of matter and how things might change when mixed, heated, pressurized, etc. Some important effects are chemical reactions (such as oxidation), formation of solutions, phase transformations.

Introduction to the Thermodynamics of Materials

Gaskell 1. Introduction to the Thermodynamics of Materials Third Edition David R. Gaskell Preliminaries † Settings Off@General::spellD † Physical Constants Needed for Problems ü Heat Capacities The generic heat capacity $C_p = a + b T + c T^2$; The heat capacities of various elements and compounds are C_p

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Thermodynamic Properties

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Another annoyance is that Gaskell's solutions in the back of the book are sometimes wrong, which means that it may be difficult to use a self-teaching book. In addition, Gaskell's solutions to some configurational entropy problems are just completely unconventional and nonsensical from an intuitive standpoint -- my teacher told us to disregard ...

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