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t. Ionic Bond. Ions pack in a crystal lattice structure. Lattice Energy. Cation {+} and anion {-} create \_\_\_\_.

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REVIEW Chemical Bonding SECTION 4

SHORT ANSWER Answer the following

questions in the space provided. 1. b In

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metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile.

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bond- a mutual electrical attraction between the nuclei and the valence electrons of different atoms that binds the atoms together ; 2. ionic bond- a chemical bond that results from the electrical attraction between anions and cations ...

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Quiz, continued Class Date 6. Metals are malleable because when struck, one plane of metal atoms can slide past another plane without breaking bonds.

- b. cannot easily move out of the way.
- c. moves in a way that maximizes the repulsive forces within the

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continued 18. When a strong acid is  
titrated with a weak base, the pH of the  
solution at the equivalence point is than

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7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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REVIEW Chemical Bonding MIXED  
REVIEW SHORT ANSWER Answer the

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following questions in the space provided. 1. a.

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c. an orbital b. an electron d. a suborbital

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